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Synthesis of New Pyrazoline Derivative & Its Antimicrobial Activity

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ABSTRACT

Some novel series of pyrazoline derivatives were synthesized from Chalcones. Chalcones were prepared by treatment of Paracetamol with different aldehyde by Claisen-Schmidt Condensation. Various Pyrazoline derivatives were prepared by reflux reaction of Chalcone with Phenyl Hydrazine/Hydrazine Hydrate in ethanolic solution. The structures of the newly synthesized Pyrazoline derivatives have been established on the basis of their spectral data. Structures of compounds were confirmed by IR, ¹H NMR. The synthesized selected compounds were screened for their antimicrobial activity.

Keywords: Pyrazoline, Chalcones, Antimicrobial activity.

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INTRODUCTION

Due to the rapid development of bacterial resistance antibacterial agents, it is vital to discover novel scaffolds for the design and synthesis of the new antibacterial agents to help in the battle against pathogenic microorganisms¹. The recent success of pyrazole COX-2 inhibitor has further highlighted the importance of these heterocycles in medicinal chemistry. A systematic investigation of this class of heterocyclic lead revealed that pyrazole containing pharmacologically active agents play an important role in medicinal chemistry. Chalcones represent an essential group of natural as well as synthetic products². Pyrazoline derivatives have displayed a wide range of biological and pharmacological activities as analgesic, anti-inflammatory, antipyretic, antiparasitic, antimalarial, antifungal, antimicrobial³. The presence of reactive α , β -unsaturated keto group in chalcones is found to be responsible for their biological activity. In the present work chalcones have been prepared according to Claisen-Schmidt condensation by condensing various ketones with aromatic aldehyde⁴⁻⁵.

The prevalence of pyrazole cores in biologically active molecules has stimulated the need for elegant and efficient ways to make these heterocyclic leads. The treatment of pain continues to be the subject of considerable pharmaceutical and clinical research. Microbial infections often produce pain and inflammation. Chemotherapeutic, analgesic and anti-inflammatory drugs are prescribed simultaneously in normal practice⁶⁻⁷.

MATERIALS AND METHOD

Procedure:

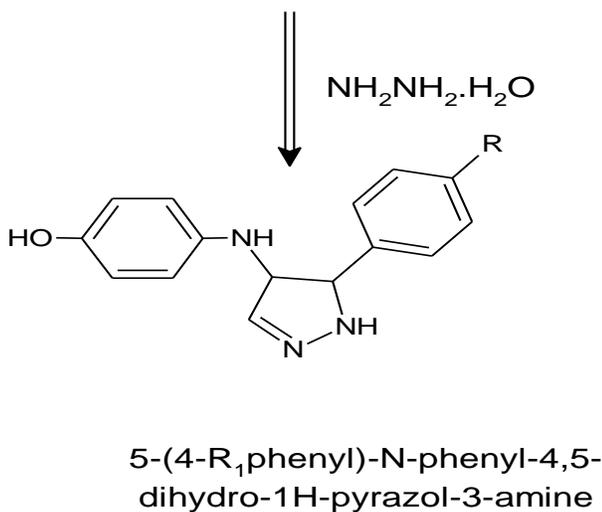
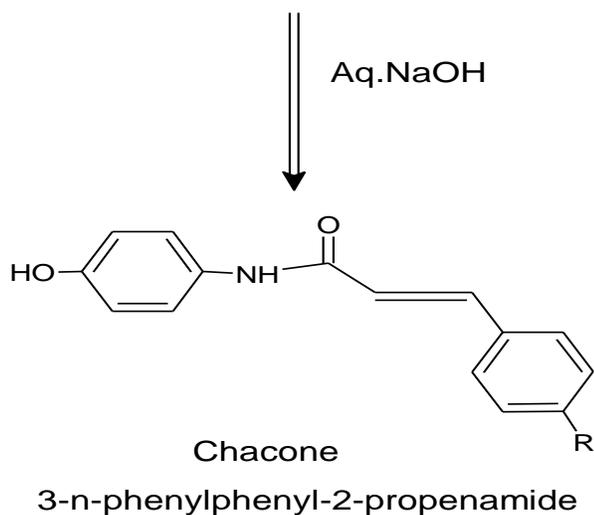
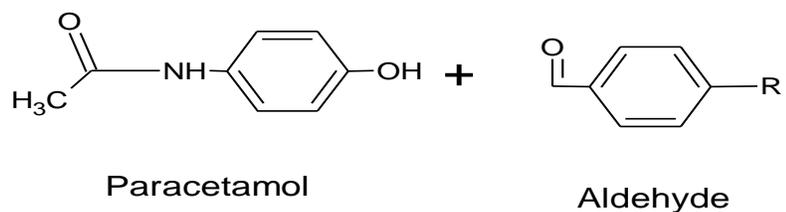
General procedure for synthesis of chalcones⁸:-

A solution of 22g of sodium hydroxide in 200ml of water and 100g (122.5ml) of rectified spirit was kept in a bath of crushed ice, (0.43mol) of freshly distilled aromatic ketone was added with stirring, and then (0.43mol) of pure aromatic aldehyde was added drop wise with stirring, temperature of the mixture was maintained at about 25°C and stirred vigorously until the mixture is so thick that stirring is no longer effective (2-3 hrs), reaction mixture was removed and kept in the refrigerator overnight. The product was filtered, washed with cold water until the washings are neutral to litmus. The crude chalcone, after drying in the air was recrystallized from rectified spirit.

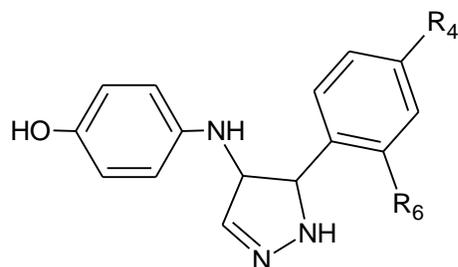
General procedure for synthesis of Pyrazoline derivatives⁹:-

A mixture of chalcone (5mmoles) and hydrazine hydrate (5mmoles) was dissolved in absolute alcohol (25ml) and refluxed for 9-10 hrs. The reaction mixture was poured onto crushed ice and

stirred, the solid thus obtained was filtered off and washed with water and taken for the next step immediately.



REACTION:-



5-(4-R phenyl)-N-phenyl-4, 5-dihydro-1H-pyrazol-3-amine

Table 1: Observation table

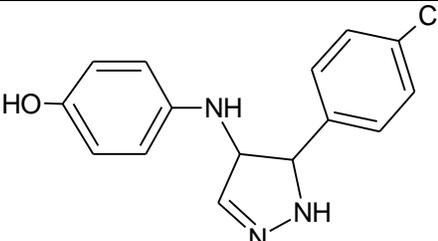
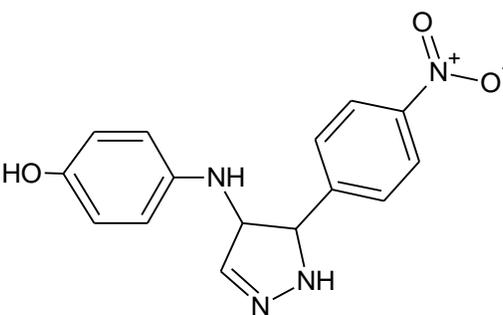
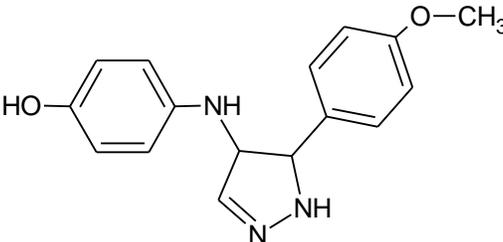
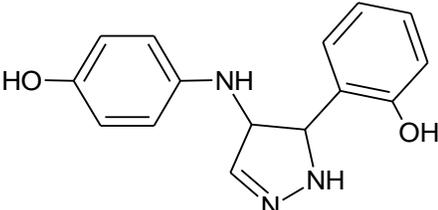
R ₄	R ₆	Structure	% yield	Molecular formula	Molecular weight	Melting point(°c)
Cl	H	 <p>5-(4-chlorophenyl)-N-phenyl-4,5-dihydro-1H-pyrazol-3-amine</p>	78%	C ₁₅ H ₁₃ N ₃ OCl	287	190-192
NO ₂	H	 <p>5-(4-nitrophenyl)-N-phenyl-4,5-dihydro-1H-pyrazol-3-amine</p>	80%	C ₁₅ H ₁₃ O ₃ N ₄	297	182-184
OCH ₃	H	 <p>5-(4-methoxyphenyl)-N-phenyl-4,5-dihydro-1H-pyrazol-3-amine</p>	75%	C ₁₆ H ₁₆ N ₃ O ₂	282	143-145
H	OH		66%	C ₁₅ H ₁₄ O ₂ N ₃	268	164-166

Table 2: Spectroscopic Evaluation

Name of compound	Functional Group	Value
3-(4-chlorophenyl)-n-phenyl acrylamide Chalcone	Ar. C-H (str.)	3053.32
	C=O	1685.79
	C=C	1583.56
	C-Cl	704.02
	C-N	1286.52
5-(4-chlorophenyl)-N-phenyl-4,5-dihydro-1H-pyrazol-3-amine	Ar. C-H (str.)	3045.6
	C=C	1581.63
	C-Cl	709.04
	N-H	3350.35
	C-N	1284.59
3-(4-nitrophenyl)-N-phenylacrylamide (Chalcone)	Ar. C-H (str.)	3051.39
	C=O	1602.85
	C=C	1602.85
	Ar.-NO ₂	1529.55
	C-N	1340.53
5-(4-Nitrophenyl)-N-phenyl-4, 5-dihydro-1H-pyrazol-3-amine	Ar. C-H (str.)	3105.39
	C=C	1595.13
	Ar-NO ₂	1517.98
	N-H	3388.93
	C-N	1334.74
5-(2-hydroxyphenyl)-N-phenyl-4, 5-dihydro-1H-pyrazol-3-amine	Ar. C-H (str.)	3045.6
	C=C	1570.06
	-OH	2945.3
	N-H	3352.28
	C-N	1327.03

NMR spectroscopy:**Table 3: 5-(4-Nitrophenyl)-N-phenyl-4,5-dihydro-1H-pyrazol-3-amine**

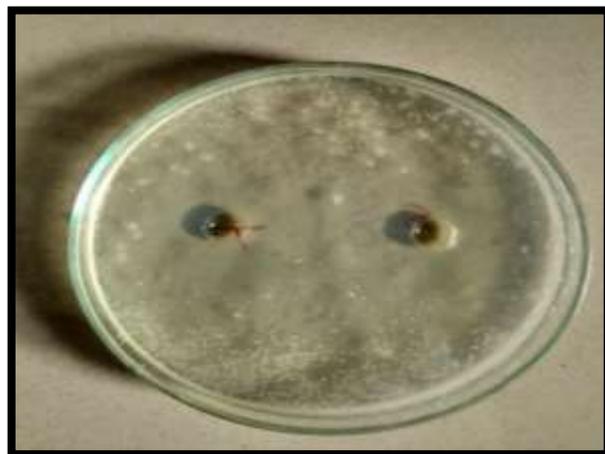
Sr.No	Spectral Data	δ ppm
1	-CH ₂	2.93
2	-Ar -CH	4.00
3	-NH	5.615
4	-Ar - Benzene	8.179

Antimicrobial activity:-

5-(4-chlorophenyl)-N-phenyl-4,5-dihydro-1H-pyrazol-3-amine



5-(4-Nitrophenyl)-N-phenyl-4,5-dihydro-1H-pyrazol-3-amine



5-(2-hydroxyphenyl)-N-phenyl-4,5-dihydro-1H-pyrazol-3-amine

Figure 1: Photographs of antimicrobial activity

Table 4: Antimicrobial activity data of pyrazoline derivatives is represented in table

Plate no.	Compounds	Zone of inhibition (mm)	
		Test	Standard(Ciprofloxacin)
1	5-(4-chlorophenyl)-N-phenyl-4,5-dihydro-1H-pyrazol-3-amine	21.5	19.2
2	5-(4-Nitrophenyl)-N-phenyl-4,5-dihydro-1H-pyrazol-3-amine	25	24.3
3	5-(4-hydroxyphenyl)-N-phenyl-4,5-dihydro-1H-pyrazol-3-amine	24.5	19.8

Table 5: Mobile phase used for TLC is Toluene: Acetone: Ammonia (7:3:0.1)

Sr. No.	Name of compounds	Rf Value
1	5-(4-chlorophenyl)-N-phenyl-4,5-dihydro-1H-pyrazol-3-amine	0.62
2	5-(4-Nitrophenyl)-N-phenyl-4,5-dihydro-1H-pyrazol-3-amine	0.74

3	5-(4-methoxyphenyl)-N-phenyl-4,5-dihydro-1H-pyrazol-3-amine	0.59
4	5-(4-hydroxyphenyl)-N-phenyl-4,5-dihydro-1H-pyrazol-3-amine	0.60

RESULT AND DISCUSSION

The compounds were synthesized according to the procedures as given in the experimental section. The purpose of the present study was to examine whether molecular modification might result in detection of new potential antimicrobial drugs. A series of compounds were prepared and assayed in a biological test for antimicrobial activity. The compounds were further characterized by IR and H^1 NMR. All synthesized pyrazoline derivatives shows antimicrobial activity. These compounds are found to be active against staphylococcus aureus (Gram-positive).

CONCLUSION

In conclusion, the results of this investigation revealed that the observed increase in antimicrobial activities are attributed to the presence of 4-NO₂, 4-Cl, 4-OCH₃, and 2-OH in phenyl ring at 5-position of pyrazoline ring of synthesized compounds. Obviously, the comparative evaluation of active compounds will required further studies; the data reported in this article may be helpful guide for the medicinal chemist who are working in this area.

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