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## Synthesis and Antimicrobial Activity of 1-Aryl-5-Hepta-O-Benzoyl-B-D-Lactosyl-2-S-Benzyl-2, 4-Isodithiobiurets

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### ABSTRACT

Several 1-aryl-5-hepta-O-benzoyl- $\beta$ -D-lactosyl-2-S-benzoyl-2, 4-isodithiobiurets have been synthesized by the interaction of hepta-O-benzoyl- $\beta$ -D-lactosyl isothiocyanate with several 1-aryl-S-benzyl isothiocarbamides. These compounds were screened for their antibacterial and antifungal activities against *Escherichia coli*, *Proteus vulgaris*, *Staphylococcus aureus*, *Salmonella typhimurium*, *Klebsiella pneumoniae*, *Pseudomonas aeruginosa*, *Aspergillus niger* and *Candida albicans*. The newly synthesized compounds have been characterized by analytical and IR, <sup>1</sup>H NMR and Mass spectral studies. These compounds show appreciable activity towards these microorganisms.

**Keywords:** Lactosyl isothiocyanate, 1-aryl-S-benzyl isothiocarbamides, isodithiobiurets and antimicrobial activity.

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## INTRODUCTION

Aryl/alkyl isothiocarbamides, due to their basic nature are found to interact with isothiocyanate to form corresponding isodithiobiurests. Several non-glycosidic isodithiobiurests are known for their anticonvulsant and hypnotic activities, Glycobiology<sup>1</sup> has gained much attention because the oligosaccharide part and other glycoconjugates are responsible for their function in various biological processes viz. cell growth. Regulation, immunological responses, inflammation and bacterial and viral infections<sup>2-4</sup>. Literature survey reveals that synthesis of amino, diamino derivatives which exhibit biological and pharmaceutical activities such antimalarial effect<sup>5, 6</sup>. Glycosyl thiourea has been widely used as important intermediate in the synthesis of nucleoside analogs<sup>7-9</sup>. Thiobiurets, imidazoles and thiazolines also shows anti inflammatory, antitumor, hypnotic activities<sup>10,11</sup>. In recent years, steadily increasing research effort has centered on the production of glycosylbiurets because these compounds have been shown to possess many different biological activities. Some carbohydrate base urea exhibit relevant biological properties such as the antibiotic SF-1993, CV-1. Nitroso urea have shown to be alpha-glycosidase inhibitors, possesses antitumor activity. In the last years the intensive use of antibiotic has lead to an increase of the emergence of resistant bacteria.<sup>12</sup>. There is a growing need for new class of antibacterial compounds having different mechanism of action compared to existing drugs.

## MATERIALS AND METHOD

### Experimental

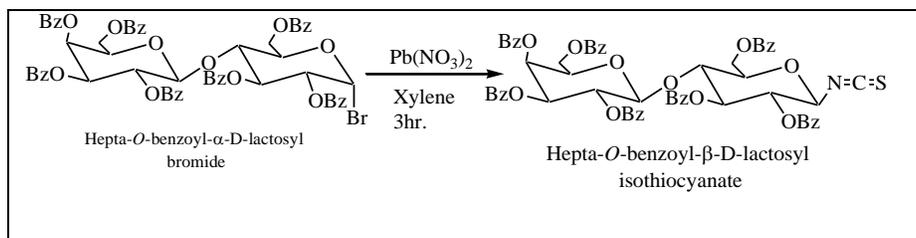
Melting points of all synthesized compounds were determined using open capillary tube on Mac digital melting point apparatus and were uncorrected. The reagent grade chemicals were obtained from commercial sources and purified by either distillation or recrystallization before use. IR spectra were recorded in solid phase KBr disks on SHIMADZU IR affinity-1 FTIR spectrometer and <sup>1</sup>H NMR spectra in CDCl<sub>3</sub> on Bruker DRX-300 of NMR spectrometer 300 MHz. The Mass spectra were recorded on Waters UPLC-TQD Mass Spectrometer. Optical rotations were measured on Equip-Tronics EQ 800 Digital Polarimeter in CHCl<sub>3</sub>. Purity of synthesized compounds has been checked by thin layer chromatography. It was performed on E. Merck pre-coated silica gel plates.

### General Procedure

#### Preparation of hepta-*O*-benzoyl- $\beta$ -D-lactosylisothiocyanate

To a suspension of hepta-*O*-benzoyl- $\alpha$ -D-lactosyl bromide (0.01 M, 15 g) in sodium dried xylene (80 ml) was added lead thiocyanate (0.01 M, 4.2 g). The reaction mixture was refluxed gently for 3 hr. With frequent shaking. This solution was then cooled and librated lead bromide was removed

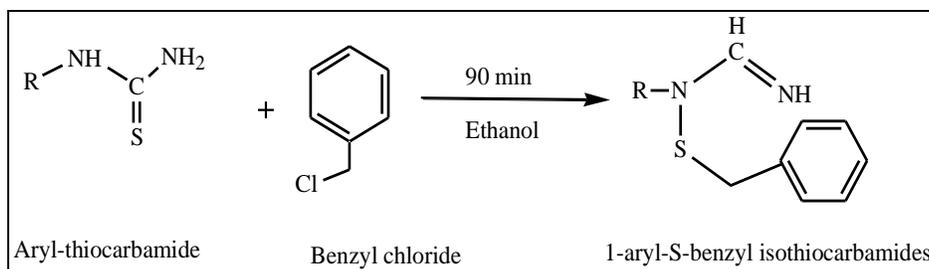
by filtration. The xylene filtrate was then treated with petroleum ether (60 – 80°C) with steering a white solid mass obtained (13 g). This solid was expected hepta-*O*-benzoyl- $\beta$ -D-lactosylisothiocyanate(I). It was purified by dissolving it in minimum quantity of chloroform and reprecipitating with petroleum ether. m.p.-118-120°C. (I) (Scheme-I).



**Figure 1: Synthesis of Hepta-*O*-benzoyl- $\beta$ -D-lactosylisothiocyanate**

### Preparation of 1-aryl-*S*-benzyl isothiocarbamides

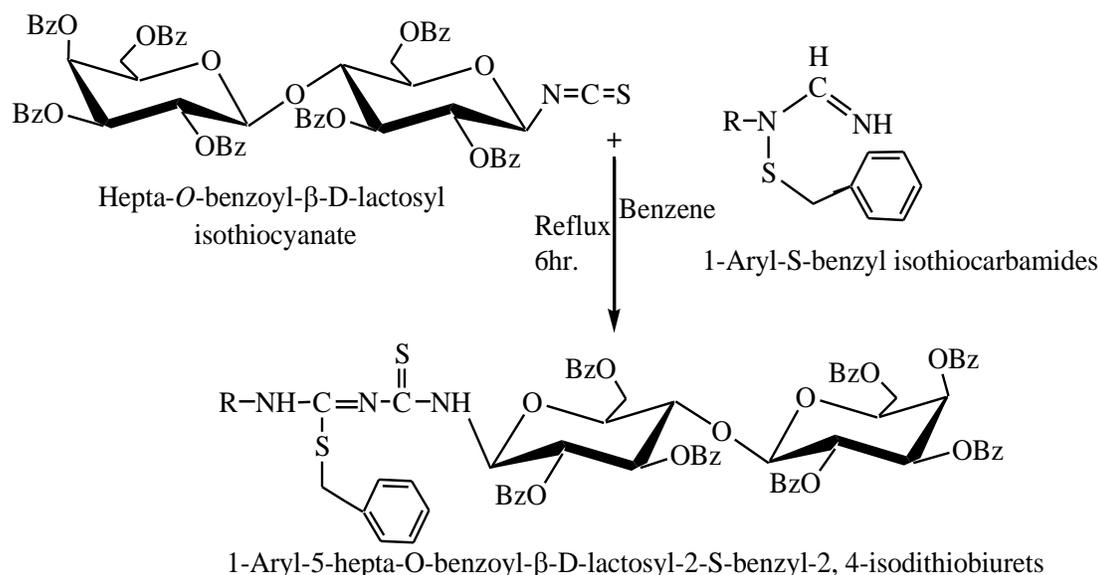
The required 1-aryl-*S*-benzyl isothiocarbamides were prepared by already well known procedure. A solution of hydroxy phenyl thiocarbamide (0.02 M, 3.36) in ethanol (25 ml) was mixed with benzyl chloride (0.02 m, 2.35 g) in ethanol (25 ml) and the reaction mixture was refluxed for 90 min. afterwards, it was cooled and rendered basic with dilute ice cold ammonia solution as a sticky residue was obtained which on standing for 2 hr., solidified then it was filtered and washed with petroleum ether. This solid was expected to be 1-aryl-*S*-benzyl isothiocarbamides II (a-f) (Scheme-II).



**Figure 2: Synthesis of 1-aryl-*S*-benzyl-isothiocarbamide**

### Preparation of 1-*o*-hydroxy-phenyl-5-hepta-*O*-benzoyl- $\beta$ -D-lactosyl-2-*S*-benzyl-2, 4-isodithiobiurete

A benzene solution of 1-*o*-hydroxy phenyl-*S*-benzyl isothiocarbamide (0.002 M, 0.518g in 20 mL) was added to a benzene solution of hepta-*O*-benzoyl-1- $\beta$ -D-lactosyl isothiocyanate (0.002 M, 2.22 g in 20 mL). The reaction mixture was refluxed for 6 hr and monitored by TLC afterwards, solvent was removed under reduced pressure to obtain sticky residue. This was triturated with petroleum ether (60 – 80°C) to afforded a pale yellow solid (88.99%). The crude product was 1.21 g crystallized by ethanol water m.p. 77°C, Anal. Calcd. For C<sub>76</sub>H<sub>64</sub>O<sub>18</sub>N<sub>3</sub>S<sub>2</sub>, Required: C, 66.61; H, 4.67; N, 3.06; S, 4.67: III (a-f) (Scheme-III).



**Figure 3: 1-aryl-5-hepta-*O*-benzoyl- $\beta$ -D-lactosyl-2-S-benzyl-2, 4-isodithiobiurets**

Where, OBz = Benzoyl

R = a) *o*-hydroxyphenyl b) *m*-hydroxyphenyl c) *p*-hydroxyphenyl

d) *o*-methoxy phenyl e) *m*-methoxy phenyl f) *p*-methoxy phenyl

## RESULTS AND DISCUSSION

Herein, we report the synthesis of various 1-hydroxy phenyl-5-hepta-*O*-benzoyl- $\beta$ -D-lactosyl-2-S-benzoyl-2, 4-isodithiobiurets III(a-g) have been synthesized by the interaction of hepta-*O*-benzoyl- $\beta$ -D-lactosylisothiocyanate (I) with several 1-aryl-S-benzyl isothiocarbamides II(a-g) in benzene medium. All products were crystallized from ethanol before recording the physical data (Table-1). The purity of compounds was checked by TLC. The spectral analysis<sup>13-15</sup> IR, <sup>1</sup>H NMR and Mass spectra of the product were observed. Optical rotation of the product was also recorded III (a-g) (Scheme-II).

**Table 1: Physical characterization of 1-aryl-5-hepta-*O*-benzoyl- $\beta$ -D-lactosyl-2-S-benzoyl-2, 4-isodithiobiurets (IIIa-f) (Scheme III)**

Sr. No	Comp.	Yield%	m. p. °C	Elemental analysis Found (Required)		[ $\alpha$ ] <sub>D</sub> <sup>28</sup> (c, CHCl <sub>3</sub> )	R <sub>f</sub> Petroleum ether:EtoAc,7: 3)
				N	S		
1	IIIa	64%	77°C	3.08(3.06)	4.65 (4.67)	+22.55°,0.115	0.87
2	IIIb	82 %	120°C	3.05(3.06)	4.62 (4.67)	+57.59°,0.111	0.75
3	IIIc	85 %	107°C	3.10(3.06)	4.70 (4.67)	+102.9°,0.132	0.50
4	III d	76.5%	128-130	3.08(3.03)	4.66 (4.62)	-128°, 0.112	0.65
5	IIIe	80%	132	3.05(3.03)	4.65 (4.62)	+36.8°, 0.133	0.48
6	III f	65%	158-160	2.98(3.03)	4.58 (4.62)	-87.23, 0.143	0.58

**Reactants:** i) **Hepta-*O*-benzoyl- $\beta$ -D-lactosylisothiocyanate,** ii) **1-aryl-S-benzyl isothiocarbamids**

### Spectral Data

**IIIa) 1-*o*-hydroxy-phenyl-5-hepta-*O*-benzoyl- $\beta$ -D-lactosyl-2-*S*-benzyl-2, 4-isodithiobiuret:**  
m.p.: 77°C; Yield : 64.00%;  $[\alpha]_D^{28}$  : + 22.55° (c,0.115 in CHCl<sub>3</sub>);

**IR (KBr, cm-1):**  $\nu$ , 3458 (N-H stretch), 3061 (Ar-H stretch), 2954 (Ali C-H stretch)1730 (C=O), 1492 (C=N), 1313 (C-N), 1379 (C-O), 1026 and 910 (characteristic of lactose), 746 (C-S);

**<sup>1</sup>H NMR (CDCl<sub>3</sub>, ppm):**  $\delta$ 8.11- 7.18 ppm(44H, m, Ar-H), 6.70 -6.60(2H, m, NH),  $\delta$ 6.22 – 3.69 (14H, m, lactosyl protons);

**Mass (m/z):** (M<sup>+</sup>)-1369,(M<sup>+</sup>-OH) -1352, (HBL<sup>+</sup>)-1053, (HBL<sup>+</sup>-C<sub>7</sub>H<sub>5</sub>O)-948, (HBL<sup>+</sup>-C<sub>7</sub>H<sub>5</sub>O)-948, (HBL<sup>+</sup>-C<sub>5</sub>H<sub>5</sub>O<sub>2</sub>) -931, (TBG)-579, (TBG-C<sub>6</sub>H<sub>5</sub>)-474, (C<sub>6</sub>H<sub>5</sub>O<sub>2</sub>)-109,C<sub>7</sub>H<sub>7</sub>-91,. (Anal.Calcd. For C<sub>76</sub>H<sub>64</sub>O<sub>18</sub>N<sub>3</sub>S<sub>2</sub>, Required: C, 66.61; H, 4.67; N, 3.06; S, 4.67 Found: C, 66.65; H, 4.70; N, 3.08; S, 4.65 %).

**IIIb) 1-*m*-hydroxy-phenyl-5-hepta-*O*-benzoyl- $\beta$ -D-lactosyl-2-*S*-benzyl-2, 4-isodithiobiuret:**  
m.p. :120°C; Yield : 82.00%;  $[\alpha]_D^{28}$  : + 57.59° (c,0.111 in CHCl<sub>3</sub>);

**IR (KBr, cm-1):**  $\nu$ , 3352 (N-H stretch), 3061 (Ar-H stretch), 2916 (Ali C-H stretch)1730 (C=O), 1492 (C=N), 1315 (C-N), 1379 (C-O), 1026 and 910 (characteristic of lactose), 709 (C-S);

**IIIc) 1-*p*-hydroxy-phenyl-5-hepta-*O*-benzoyl- $\beta$ -D-lactosyl-2-*S*-benzyl-2, 4-isodithiobiuret:**  
m.p.:107°C; Yield: 85.00%;  $[\alpha]_D^{28}$ : + 102.9° (c, 0.132 in CHCl<sub>3</sub>);

**IR (KBr, cm-1):**  $\nu$ , 3462 (N-H stretch), 3061 (Ar-H stretch), 2976 (Ali C-H stretch) 1730 (C=O), 1492 (C=N), 1315 (C-N), 1379 (C-O), 1070 and 908 (characteristic of lactose), 707 (C-S);

**IIId) 1-*o*-methoxy-phenyl-5-hepta-*O*-benzoyl- $\beta$ -D-lactosyl-2-*S*-benzyl-2, 4-isodithiobiuret:**  
m.p. :128-130°C; Yield : 76.50%;  $[\alpha]_D^{28}$  : -128° (c,0.112 in CHCl<sub>3</sub>);

**IR (KBr, cm-1):**  $\nu$ , 3462 (N-H stretch), 3062 (Ar-H stretch), 2976 (Ali C-H stretch)1730 (C=O), 1452 (C=N), 1315 (C-N), 1379 (C-O), 1026 and 916 (characteristic of lactose), 711 (C-S);

**<sup>1</sup>H NMR (CDCl<sub>3</sub>, ppm):**  $\delta$ 8.133- 6.221 ppm (44H, m, Ar-H),  $\delta$  6.19 (2H, m, NH),  $\delta$ 6.221 –4.587 (14H, m, lactosyl protons),  $\delta$ 3.870 (3H, s CH<sub>3</sub>);

**Mass (m/z):** (M<sup>+</sup>)-1383, (M<sup>+</sup>-OCH<sub>3</sub>)-1352, (M<sup>+</sup>-C<sub>7</sub>H<sub>8</sub>ON)-1261, (HBL<sup>+</sup>)-1053, (HBL-C<sub>5</sub>H<sub>8</sub>)-984, (HBL-C<sub>14</sub>H<sub>15</sub>O<sub>7</sub>)-757, (TBG)-579, (TBG-C<sub>7</sub>H<sub>8</sub>O<sub>2</sub>)-455, (TBG-C<sub>5</sub>H<sub>10</sub>O<sub>6</sub>)-413, (C<sub>12</sub>H<sub>22</sub>O<sub>11</sub>)-342, (TBG-C<sub>13</sub>H<sub>22</sub>O<sub>8</sub>)-273 (100%), (C<sub>8</sub>H<sub>4</sub>N<sub>3</sub>S<sub>2</sub>)-206, (C<sub>7</sub>H<sub>5</sub>N<sub>2</sub>S)-149. (Anal. Calcd. For C<sub>77</sub>H<sub>66</sub>O<sub>18</sub>N<sub>3</sub>S<sub>2</sub>Required: C, 66.81; H, 4.77; N, 3.03; S, 4.62 Found: C, 66.85; H, 4.73; N, 3.08; S, 4.66 %).

**IIIe) 1-*m*-methoxy-phenyl-5-hepta-*O*-benzoyl- $\beta$ -D-lactosyl-2-*S*-benzyl-2, 4-isodithiobiuret:**  
m.p.:132°C; Yield: 80.00%;  $[\alpha]_D^{28}$ : + 36.8° (c,0.133 in CHCl<sub>3</sub>);

**III f) 1-*p*-methoxy-phenyl-5-hepta-*O*-benzoyl- $\beta$ -D-lactosyl-2-*S*-benzyl-2, 4-isodithiobiuret:**  
m.p.:158-160°C; Yield: 65.00%;  $[\alpha]_D^{28}$ : -87.23° (c, 0.143 in CHCl<sub>3</sub>);

### Antimicrobial Studies

#### Antibacterial Activity

All the compounds have been screened for antimicrobial activity by using disc diffusion assay. The disc bearing plates were incubated at 37°C for 24 hrs. Inhibition zones read after incubation at 37°C for 24 hr. for bacterial strains. The compounds were taken at a concentration or 1mg/ml using dimethyl sulphoxide as a solvent. Amikacin (100 ug/ml) was used as standard for antibacterial. These compounds were screened for antibacterial activity against *Escherichia coli*, *Proteus vulgaris*, *Staphylococcus aureus*, *Salmonella typhi*, *Klebsiellapneumonie*, *Pseudomonasaeruginosa* in nutrient agar medium.

#### Antifungal Activity

All the compounds were also screened for their antifungal activities by using disc diffusion assay. These compounds were taken at a concentration or 1mg/ml using dimethyl sulphoxide as a solvent. Fluconazole (100ug/ml) was used as standard for antifungal activity. The results are presented in Table 2.

**Table: 2 Antimicrobial activities of of newly synthesized1-aryl-5-hepta-*O*-benzoyl- $\beta$ -D-lactosyl-2-*S*-benzoyl-2, 4-isodithiobiurets (IIIa-f).**

Compounds	<i>E. coli</i>	<i>S. aureus</i>	<i>P. vulgaris</i>	<i>Ps. aeruginosa</i>	<i>S. typhi</i>	<i>K.pneumonie</i>	<i>A. niger</i>	<i>C.albicans</i>
IIIa	11	12	10	12	15	11	20	18
IIIb	10	16	11	14	17	-	18	20
IIIc	10	12	13	13	19	-	-	-
III d	15	18	14	13	14	13	12	-
IIIe	13	19	17	10	15	10	22	15
III f	12	-	13	-	16	-	17	22
DMSO	-	-	-	-	-	-	-	-
Amikacin	18	21	23	19	20	20	-	-
Fluconazole	-	-	-	-	-	-	24	24

Zone size was interpreted by

Sample	Disc content	Resistant	Intermediate	Sensitive
Amikacin	100ug/ml	≤ 15 mm	16-20 mm	≥ 21 mm
Fluconazole	100ug/ml	≤ 15 mm	16-20 mm	≥ 21 mm

## CONCLUSION

In this research work, the characterizations of newly synthesized products were established on the basis of UV, IR, <sup>1</sup>H NMR, & Mass spectral studies. Various 1-aryl-5-hepta-O-benzoyl-β-D-lactosyl-2-S-benzoyl-2, 4-isodithiobiurets were synthesized and yield of product ranged from 64-85%. Some of the compounds synthesized showed promising antimicrobial activities.

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## REFERENCES:

1. Korpe GV, Deshmukh SP. N-glucosylated compounds: synthesis of new 1-tetra-O-acetyl- β -D-glucopyranosyl-3-aryl thiocarbamides. *Asian J Chem* 2001; 13(2):765-767.
2. Garcia Fernandez Jose M, Mellet Carmen Ortiz. Chemistry and development of N-thiocarbonyl carbohydrate derivatives sugar isothiocyanates, thioamides, thioureas, thiocarbamates and their conjugates. *AdvCarbohychem* 2000; 55:35-135.
3. Fuentes J, Angulo M, Prodera MA. Fluoronucleosides, isothiocyanato C-nucleosides and thiourylene Di-C-nucleosides via cyclic sulfates. *J Org Chem* 2002; 67(8):2577-2587.
4. Tale PV, Deshmukh SP. Synthesis of N-lactosylatedthiourea and benzothiazolyl thiourea. *Heteroatom Chem* 2006; 17(4):306-309.
5. Tale PV, Deshmukh SP. Synthesis of 2-phenylimino-3-aryl-4-S-benzyl-6-hepta-O-acetyl- β -D-lacosylimino-2,3-dihydro-1,3,5-thiadiazine hydrochlorides. *Ind J Chem* 2006; 45B (2):558-560.
6. Ghuge RD, Deshmukh SP. Synthesis and characterization of 1-aryl-5-hepta-O-acetyl- β -D-maltosyl-2-S-benzyl-2,4-isodithiobiurets. *E J Chem* 2012; 9(1):330-334.
7. Mannuel Juan Benito, Carmen Ortiz Mellet, KashinathSadalpуре, Lindhorst K Thisbe, Jacques Defaye, Jose Manuel Garcia Fernandez. Synthesis and anomeric stability of (1, 6)-thiourea-linked pseudo oligosaccharides. *Carbohydr Res* 1999; 320: 37-48.
8. Dhonde Madhukar G, Wankhade Atul V, DeshmukhShirish P. Synthesis and biological studies of unsymmetrical mannopyranosylthiocarbamides. *J Chem Pharm Res* 2010;2(4):518-525.
9. YadgireAtul V, Korpe Gajanan V, Deshmukh Shirish P. Comparative study of Microwave Induced and conventional synthesis of acetylated sugar isothiocyanates and related

- thiocarbamides. E J Chem 2011; 8(4):1614-1619.
10. Korpe GV, Deshmukh SP, Musaddiq M. Synthesis of 5-glucosyl-2, 4-isodithiobiurets and their antimicrobial study. Asian J Chem 2002; 14(1):121-124.
  11. Mangte DV, Deshmukh SP. On lactosylthioureides: synthesis of certain S-hepta-O-acetyl-lactosyl-1-arylisothiocarbamides. J Ind Chem Soc 2005; 82:1025-1026.
  12. Werner EW. Contribution to the chemistry of thiocarbamides, interaction of benzyl chloride and allyl bromide, respectively with thiocarbamide, monophenyl thiocarbamide and dipheylthiocarbamide. J Chem Soc 1890; 57:283-294.
  13. Silverstein RM and Webster FX, "Spectrometric Identification of Organic Compounds", 6th ed., John Wiley and Sons, Inc, New York; 2011.
  14. Williams DH and Fleming I, "Spectroscopic Methods in Organic Chemistry", 5th ed., Tata McGraw-Hill; 2004.
  15. Dyer JR, "Applications of Absorption Spectroscopy of Organic Compounds", PHI Learning Private Limited, New Delhi; 2010.

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