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Solvent-Free and Conventional Approach to the Synthesis of Indazole Derivatives through Cyclohexenone and Their Antibacterial Activity

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ABSTRACT

Microwave assisted solvent free synthesis of cyclohexenone and indazole derivatives were carried out by chalcones derived from different acetophenones and 1-naphthaldehyde. Structures of the compounds have been characterized by IR, ¹H NMR, Mass spectrometry and elemental analysis. All the synthesized indazole derivatives were screened for antibacterial activity and are found to possess moderate to good activity against some species of bacteria.

Keywords: 1-naphthaldehyde, chalcones, cyclohexenone, indazole, antibacterial, microwave irradiation.

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INTRODUCTION

Indazole or benzpyrazole or isoindazone¹, is a heterocyclic aromatic compound with a rare occurrence in nature. The indazole nucleus is a pharmaceutically important and emerging heterocycle with a broad spectrum of activities. These are found to possess anticancer² anti-inflammatory^{3,4} antiproliferative^{5,6} hypotensive⁷ antitubercular⁸ protein kinase inhibition⁹, anti-angiogenic¹⁰ antimicrobial activities¹¹⁻¹³ etc. A number of interesting organic syntheses have been performed using the technique of solvent free synthesis. The solvent-free synthesis is found to be environmentally friendly and economical since it avoids the use of solvents and offers easy work-up procedure^{14,15}. Also, the yield of product is found to be more than that of conventional technique.

MATERIALS AND METHOD

The materials were obtained from commercial suppliers and were used without further purification. Melting points were determined in an open capillary tube and are uncorrected. The reactions were monitored by thin-layer chromatography (TLC) analysis using silica gel (Merck 60 F₂₅₄) plates. Compounds were visualized by UV irradiation. IR spectra were recorded in KBr on a Perkin-Elmer spectrometer (RX-IFTIR). ¹H NMR spectra were recorded on a Bruker Avance II 400 NMR spectrometer in DMSO as solvent and TMS as an internal standard. Mass spectra were recorded on Advion mass spectrometer by ESI source. Elemental analysis was carried out at Thermo Scientific (FLASH 2000) CHN elemental analyzer. Microwave irradiation was carried out in a CEM Discover scientific microwave oven.

Synthesis of Chalcones(3a-g)

Various aromatic acetophenones (0.01 mol) in ethanol 30 ml and 1-naphthaldehyde (0.01 mol) were taken and the solution was heated to near boiling. Aqueous 40% NaOH solution (0.03 mol, 3 ml) was added to it dropwise with constant stirring. The mixture was further stirred at room temperature to yield solid mass. The reaction mixture was kept overnight. Then it was poured over crushed ice and acidified with conc. HCl. The solid thus obtained was filtered, washed with water and purified by crystallization using proper solvent.

Process I (Conventional Method)

Synthesis of Cyclohexenone derivatives (4a-g)

A mixture of chalcone (0.01 mol), ethyl acetoacetate (0.01 mol) and 0.5 ml 10% NaOH in 20 ml ethanol was refluxed for 1-5 hrs. After completion of reaction as indicated by TLC, the product was isolated by extracting with diethyl ether. The ether phase was separated and dried over

anhydrous Na_2SO_4 prior to be subjected to solvent removal. The solid residue obtained was recrystallized from aqueous ethanol.

Synthesis of Indazole derivatives (5a-g)

To a solution of compound (4a-g) (0.01 mol) in absolute ethanol (10 ml) containing glacial acetic acid (0.5 ml), hydrazine hydrate (0.02 mol) was added. The reaction mixture was refluxed for 2-5 hrs. After the completion of reaction as indicated by TLC, the reaction mixture was cooled and poured over crushed ice with constant stirring. The solid formed was filtered off, dried and recrystallized from ethanol.

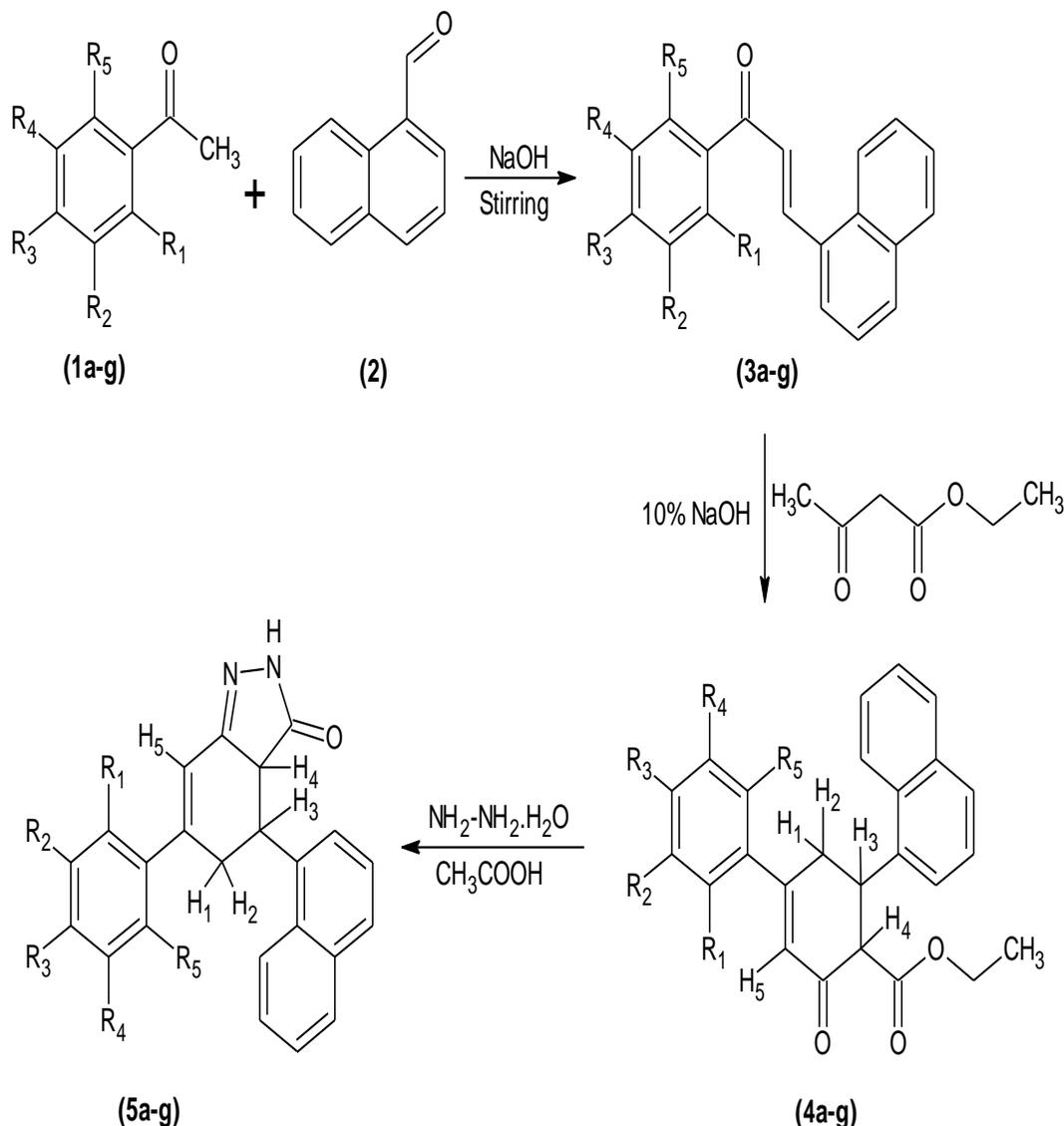


Figure-1: Synthesis of Cyclohexenone and Indazole Derivatives

Where

4a, 5a: $\text{R}_1, \text{R}_2, \text{R}_3, \text{R}_4, \text{R}_5 = \text{H}$

4b, 5b: $\text{R}_1 = \text{OH}; \text{R}_2, \text{R}_3, \text{R}_4, \text{R}_5 = \text{H}$

4c, 5c: R₁, R₂, R₄, R₅=H; R₃=OH

4d, 5d: R₁=OH; R₂, R₃, R₅=H; R₄=CH₃

4e, 5e: R₁=OH; R₂, R₃, R₅=H; R₄=Cl

4f, 5f: R₁=OH; R₂, R₃, R₅=H; R₄=Br

4g, 5g: R₁=OH; R₂, R₄=Cl; R₃, R₅=H

Process II (Microwave method)

Synthesis of Cyclohexenone derivatives (4a-g)

A mixture of chalcone (0.01 mol), ethyl acetoacetate (0.01 mol) and 0.5 ml 10% NaOH were taken in a 100 ml RB flask and irradiated in a microwave oven at 300 Watts for 2.5 - 6 min with intermittent cooling after 30 sec. After the completion of reaction as indicated by TLC, cold water was added to the flask and stirred for few minutes. The product formed was filtered and dried. It was recrystallized from aqueous ethanol to afford the desired product.

Synthesis of Indazole derivatives (5a-g)

A mixture of compound (4a-g) (0.01 mol), hydrazine hydrate (0.02 mol) and glacial acetic acid (0.5 ml) was taken in a 100 ml RB flask and irradiated in a microwave oven at 300 Watts for 2 - 5.5 min with intermittent cooling after 30 sec. After the completion of reaction as indicated by TLC, cold water was added to the flask and stirred for few minutes. The product formed was filtered and dried. It was recrystallized from ethanol to afford the desired product.

Ethyl 6-(naphthalen-1-yl)-2-oxo-4-phenylcyclohex-3-enecarboxylate (4a):

MS: [M+H]⁺ 371; IR (KBr) cm⁻¹: 1728 (C=O ester), 1630 (C=O cyclic); ¹H NMR (δ, DMSO): δ 0.84-0.88 (3H, t, J= 7.2 Hz, -CH₂-CH₃), 2.50-2.53 (2H, m, H₁ & H₂), 3.06-3.08 (1H, d, J= 8 Hz, H₄), 3.20-3.22 (1H, m, H₃), 4.55-4.60 (2H, q, -CH₂-CH₃), 6.58 (1H, s, H₅), 7.08- 8.22 (12H, m, Ar-H); Anal.Calcd for C₂₅H₂₂O₃: C, 81.06; H, 5.99. Found: C, 81.15; H, 5.80%.

Ethyl 4-(2-hydroxyphenyl)-6-(naphthalen-1-yl)-2-oxocyclohex-3-enecarboxylate (4b):

MS: [M+H]⁺ 387; IR (KBr) cm⁻¹: 1732 (C=O ester), 1636 (C=O cyclic); ¹H NMR (δ, DMSO): δ 0.85-0.90 (3H, t, J = 8 Hz, -CH₂-CH₃), 2.50-2.51 (2H, m, H₁ & H₂), 3.03-3.05 (1H, d, J = 8 Hz, H₄), 3.15-3.21 (1H, m, H₃), 4.58-4.61 (2H, q, -CH₂-CH₃), 6.43 (1H, s, H₅), 6.43-8.22 (11H, m, Ar-H), 9.98 (1H, s, OH); Anal.Calcd for C₂₅H₂₂O₄: C, 77.70; H, 5.74. Found: C, 77.84; H, 5.69%.

Ethyl 4-(4-hydroxyphenyl)-6-(naphthalen-1-yl)-2-oxocyclohex-3-enecarboxylate (4c):

MS: [M+H]⁺ 387; IR (KBr) cm⁻¹: 1725 (C=O ester), 1628 (C=O cyclic); ¹H NMR (δ, DMSO): δ 0.84-0.87 (3H, t, J= 7.4 Hz, -CH₂-CH₃), 2.47-2.52 (2H, m, H₁ & H₂), 3.01-3.03 (1H, d, J= 7.8 Hz, H₄), 3.17-3.19 (1H, m, H₃), 4.50-4.53 (2H, q, -CH₂-CH₃), 6.54 (1H, s, H₅), 6.50-8.34 (11H, m, Ar-H), 9.87 (1H, s, OH); Anal.Calcd for C₂₅H₂₂O₄: C, 77.70; H, 5.74. Found: C, 77.62; H, 5.69%.

Ethyl4-(2-hydroxy-5-methylphenyl)-6-(naphthalen-1-yl)-2-oxocyclohex-3-enecarboxylate**(4d):**

MS: $[M+H]^+$ 401; IR (KBr) cm^{-1} : 1734 (C=O ester), 1642 (C=O cyclic); ^1H NMR (δ , DMSO): δ 0.80-0.86 (3H, t, $J=6.9$ Hz, $-\text{CH}_2-\text{CH}_3$), 2.36 (3H, s, $-\text{CH}_3$), 2.38-2.47 (2H, m, H_1 & H_2), 3.04-3.06 (1H, d, $J=7.9$ Hz, H_4), 3.22-3.24 (1H, m, H_3), 4.48-4.52 (2H, q, $-\text{CH}_2-\text{CH}_3$), 6.40 (1H, s, H_5), 6.87-7.98 (10H, m, Ar-H), 9.60 (1H, s, OH); Anal.Calcd for $\text{C}_{26}\text{H}_{24}\text{O}_4$: C, 77.98; H, 6.04. Found: C, 78.06; H, 5.94%.

Ethyl4-(5-chloro-2-hydroxyphenyl)-6-(naphthalen-1-yl)-2-oxocyclohex-3-enecarboxylate (4e):

MS: $[M+H]^+$ 422; IR (KBr) cm^{-1} : 1748 (C=O ester), 1654 (C=O cyclic); ^1H NMR (δ , DMSO): δ 0.86-0.92 (3H, t, $J=7.6$ Hz, $-\text{CH}_2-\text{CH}_3$), 2.46-2.50 (2H, m, H_1 & H_2), 3.08-3.10 (1H, d, $J=7.8$ Hz, H_4), 3.15-3.19 (1H, m, H_3), 4.46-4.50 (2H, q, $-\text{CH}_2-\text{CH}_3$), 6.42 (1H, s, H_5), 6.64-7.89 (10H, m, Ar-H), 10.01 (1H, s, OH); Anal.Calcd for $\text{C}_{25}\text{H}_{21}\text{ClO}_4$: C, 71.34; H, 5.03. Found: C, 71.44; H, 5.16%.

Ethyl4-(5-bromo-2-hydroxyphenyl)-6-(naphthalen-1-yl)-2-oxocyclohex-3-enecarboxylate (4f):

MS: $[M+H]^+$ 466; IR (KBr) cm^{-1} : 1752 (C=O ester), 1657 (C=O cyclic); ^1H NMR (δ , DMSO): δ 0.89-0.93 (3H, t, $J=7.8$ Hz, $-\text{CH}_2-\text{CH}_3$), 2.49-2.54 (2H, m, H_1 & H_2), 3.10-3.12 (1H, d, $J=8$ Hz, H_4), 3.18-3.24 (1H, m, H_3), 4.61-4.65 (2H, q, $-\text{CH}_2-\text{CH}_3$), 6.39 (1H, s, H_5), 6.54-7.71 (10H, m, Ar-H), 10.05 (1H, s, OH); Anal.Calcd for $\text{C}_{25}\text{H}_{21}\text{BrO}_4$: C, 64.53; H, 4.55. Found: C, 64.38; H, 4.67%.

Ethyl 4-(3,5-dichloro-2-hydroxyphenyl)-6-(naphthalen-1-yl)-2-oxocyclohex-3-enecarboxylate**(4g):**

MS: $[M+H]^+$ 456; IR (KBr) cm^{-1} : 1742 (C=O ester), 1672 (C=O cyclic); ^1H NMR (δ , DMSO): δ 0.91-0.94 (3H, t, $J=6.8$ Hz, $-\text{CH}_2-\text{CH}_3$), 2.44-2.48 (2H, m, H_1 & H_2), 3.09-3.11 (1H, d, $J=7.9$ Hz, H_4), 3.19-3.22 (1H, m, H_3), 4.48-4.51 (2H, q, $-\text{CH}_2-\text{CH}_3$), 6.48 (1H, s, H_5), 6.92-8.24 (9H, m, Ar-H), 10.12 (1H, s, OH); Anal.Calcd for $\text{C}_{25}\text{H}_{20}\text{Cl}_2\text{O}_4$: C, 65.95; H, 4.43. Found: C, 65.78; H 4.56%.

4-(naphthalen-1-yl)-6-phenyl-4,5-dihydro-2H-indazol-3(3aH)-one (5a):

MS: $[M+H]^+$ 339; IR (KBr) cm^{-1} : 3300 (N-H), 1685 (C=O), 1640 (C=N); ^1H NMR (δ , DMSO): δ 2.64-2.86 (2H, m, H_1 & H_2), 4.71-4.73 (1H, d, $J=8$ Hz, H_4), 5.0-5.05 (1H, m, H_3), 6.92-7.80 (12H, m, Ar-H), 8.10 (1H, s, H_5), 8.30 (1H, bs, -NH); Anal.Calcd for $\text{C}_{23}\text{H}_{18}\text{N}_2\text{O}$: C, 81.63; H, 5.36; N, 8.28. Found: C, 81.74; H, 5.45; N, 8.15%.

6-(2-hydroxyphenyl)-4-(naphthalen-1-yl)-4,5-dihydro-2H-indazol-3(3aH)-one (5b):

MS: $[M+H]^+$ 355; IR (KBr) cm^{-1} : 3219 (N-H), 1690 (C=O), 1630 (C=N); ^1H NMR (δ , DMSO): δ 2.79-3.32 (2H, m, H_1 & H_2), 3.92-3.94 (1H, d, $J=8$ Hz, H_4), 4.97-5.00 (1H, m, H_3), 6.58-7.85 (11H, m, Ar-H), 8.15 (1H, s, H_5), 8.28 (1H, bs, -NH), 9.25 (1H, s, -OH); Anal.Calcd for $\text{C}_{23}\text{H}_{18}\text{N}_2\text{O}_2$: C, 77.95; H, 5.12; N, 7.90. Found: C, 78.03; H, 5.00; N, 8.11%.

6-(4-hydroxyphenyl)-4-(naphthalen-1-yl)-4,5-dihydro-2H-indazol-3(3aH)-one (5c):

MS: $[M+H]^+$ 355; IR (KBr) cm^{-1} : 3280 (N-H), 1680 (C=O), 1600 (C=N); ^1H NMR (δ , DMSO): δ 2.76-3.10 (2H, m, H_1 & H_2), 4.42-4.44 (1H, d, $J = 8$ Hz, H_4), 4.86-4.92 (1H, m, H_3), 6.66-7.82 (11H, m, Ar-H), 8.09 (1H, s, H_5), 8.24 (1H, bs, -NH), 8.64 (1H, s, -OH); Anal.Calcd for $\text{C}_{23}\text{H}_{18}\text{N}_2\text{O}_2$: C, 77.95; H, 5.12; N, 7.90. Found: C, 78.07; H, 5.23; N, 7.81%.

6-(2-hydroxy-5-methylphenyl)-4-(naphthalen-1-yl)-4,5-dihydro-2H-indazol-3(3aH)-one (5d):

MS: $[M+H]^+$ 369; IR (KBr) cm^{-1} : 3320 (N-H), 1650 (C=O), 1580 (C=N); ^1H NMR (δ , DMSO): δ 2.28 (3H, s, $-\text{CH}_3$), 2.72-3.20 (2H, m, H_1 & H_2), 3.99-4.01 (1H, d, $J = 7.8$ Hz, H_4), 4.99-5.05 (1H, m, H_3), 6.48-7.79 (10H, m, Ar-H), 8.18 (1H, s, H_5), 8.30 (1H, bs, -NH), 8.55 (1H, s, -OH); Anal.Calcd for $\text{C}_{24}\text{H}_{20}\text{N}_2\text{O}_2$: C, 78.24; H, 5.47; N, 7.60. Found: C, 78.11; H, 5.35; N, 7.66%.

6-(5-chloro-2-hydroxyphenyl)-4-(naphthalen-1-yl)-4,5-dihydro-2H-indazol-3(3aH)-one (5e):

MS: $[M+H]^+$ 390; IR (KBr) cm^{-1} : 3380 (N-H), 1648 (C=O), 1590 (C=N); ^1H NMR (δ , DMSO): δ 2.81-3.41 (2H, m, H_1 & H_2), 4.16-4.18 (1H, d, $J = 7.9$ Hz, H_4), 4.76-4.87 (1H, m, H_3), 6.60-7.72 (10H, m, Ar-H), 8.23 (1H, s, H_5), 8.34 (1H, bs, -NH), 9.36 (1H, s, -OH); Anal.Calcd for $\text{C}_{23}\text{H}_{17}\text{ClN}_2\text{O}_2$: C, 71.04; H, 4.41; N, 7.20. Found: C, 71.17; H, 4.55; N, 7.27%.

6-(5-bromo-2-hydroxyphenyl)-4-(naphthalen-1-yl)-4,5-dihydro-2H-indazol-3(3aH)-one (5f):

MS: $[M+H]^+$ 434; IR (KBr) cm^{-1} : 3350 (N-H), 1675 (C=O), 1600 (C=N); ^1H NMR (δ , DMSO): δ 2.85-3.38 (2H, m, H_1 & H_2), 3.88-3.90 (1H, d, $J = 8$ Hz, H_4), 4.66-4.72 (1H, m, H_3), 6.50-7.86 (10H, m, Ar-H), 8.29 (1H, s, H_5), 8.35 (1H, bs, -NH), 9.39 (1H, s, -OH); Anal.Calcd for $\text{C}_{23}\text{H}_{17}\text{BrN}_2\text{O}_2$: C, 63.75; H, 3.95; N, 6.47. Found: C, 63.66; H, 3.84; N, 6.59%.

6-(3,5-dichloro-2-hydroxyphenyl)-4-(naphthalen-1-yl)-4,5-dihydro-2H-indazol-3(3aH)-one (5g):

MS: $[M+H]^+$ 424; IR (KBr) cm^{-1} : 3370 (N-H), 1662 (C=O), 1630 (C=N); ^1H NMR (δ , DMSO): δ 2.94-3.26 (2H, m, H_1 & H_2), 4.46-4.48 (1H, d, $J = 8$ Hz, H_4), 5.02-5.06 (1H, m, H_3), 6.87-7.77 (9H, m, Ar-H), 8.35 (1H, s, H_5), 8.40 (1H, bs, -NH), 10.26 (1H, s, -OH); Anal.Calcd for $\text{C}_{23}\text{H}_{16}\text{Cl}_2\text{N}_2\text{O}_2$: C, 65.26; H, 3.81; N, 6.62. Found: C, 65.05; H, 3.88; N, 6.49%.

Antibacterial activity

All the synthesized indazole derivatives were tested for in vitro antibacterial activity against two gram positive (*S. aureus*, *B. subtilis*) and two gram negative (*E. coli*, *K. aerogenes*) bacteria by well diffusion method ¹⁶ and zone of inhibition was measured in mm. Amoxicillin was used as standard drug and DMSO as control solvent. The compounds were tested at 1mg/ml concentration. The results of the activity data is shown in table-2.

RESULTS AND DISCUSSION

Mass spectra of the synthesized compounds showed peaks at $[M+H]^+$ which confirms the molecular mass of the compounds. IR spectra of cyclohexenone derivatives revealed a sharp strong band at $1725-1752\text{ cm}^{-1}$ due to C=O of ester and another sharp strong band at $1628-1672\text{ cm}^{-1}$ is due to the presence of cyclic C=O group. IR spectra of indazole derivatives showed a band at $3219-3380\text{ cm}^{-1}$ due to the presence of N-H group in the structure. A sharp strong band at $1648-1690\text{ cm}^{-1}$ is due to the presence of C=O group in the structure. C=N group in the structure gives a band between $1580-1640\text{ cm}^{-1}$. The ^1H NMR spectra of compounds (**4a-g**) gives a triplet at δ 0.80-0.94 and a quartet at δ 4.46-4.65 due to the presence of ethyl ester moiety in the structure of cyclohexenone. The two protons H_1 & H_2 appears as multiplet at δ 2.38-2.54.

The proton H_4 appears as doublet at δ 3.01-3.12 while H_3 gives multiplet at δ 3.15-3.24. The vinylic proton H_5 appears as a singlet at higher chemical shift i.e. δ 6.39-6.58. The aromatic proton gives multiplet at δ 6.43-8.34. The -OH proton in compounds (**4b-g**) appears as singlet at δ 9.60-10.12. The ^1H NMR spectra of compounds (**5a-g**) shows the absence of a triplet and a quartet at δ around 0.8 and 4.4 respectively. The -NH signal is obtained at δ 8.22-8.42. All the synthesized indazole derivatives (**5a-g**) are found to be active against *S. aureus*, *B. subtilis*, *K. aerogenes* while all are inactive against *E.coli*.

Table-1: The physicochemical data for synthesized Cyclohexenone and Indazole derivatives by conventional (Process I) and microwave (Process II) methods.

Compound	Molecular Formula	Molecular Weight	Reaction Time		% yield		Melting Point ($^{\circ}\text{C}$)
			Process I (hr)	Process II (min)	Process I	Process II	
4a	$\text{C}_{25}\text{H}_{22}\text{O}_3$	370	4	4	61	66	120
4b	$\text{C}_{25}\text{H}_{22}\text{O}_4$	386	1.5	2.5	71	74	155
4c	$\text{C}_{25}\text{H}_{22}\text{O}_4$	386	4	3	65	69	169
4d	$\text{C}_{26}\text{H}_{24}\text{O}_4$	400	2.5	2.5	68	76	172
4e	$\text{C}_{25}\text{H}_{21}\text{ClO}_4$	421	4	4	67	72	185
4f	$\text{C}_{25}\text{H}_{21}\text{BrO}_4$	465	4.5	4.5	70	77	204
4g	$\text{C}_{25}\text{H}_{20}\text{Cl}_2\text{O}_4$	455	5	6	65	72	217
5a	$\text{C}_{23}\text{H}_{18}\text{N}_2\text{O}$	338	3	3	67	74	196
5b	$\text{C}_{23}\text{H}_{18}\text{N}_2\text{O}_2$	354	2	2	70	76	230
5c	$\text{C}_{23}\text{H}_{18}\text{N}_2\text{O}_2$	354	2.5	3.5	62	68	244
5d	$\text{C}_{24}\text{H}_{20}\text{N}_2\text{O}_2$	368	2	2.5	72	81	264
5e	$\text{C}_{23}\text{H}_{17}\text{ClN}_2\text{O}_2$	389	3.5	4	74	83	257
5f	$\text{C}_{23}\text{H}_{17}\text{BrN}_2\text{O}_2$	433	4	4.5	68	75	270
5g	$\text{C}_{23}\text{H}_{16}\text{Cl}_2\text{N}_2\text{O}_2$	423	5	5.5	60	69	276

Table-2: Antibacterial activity of indazole derivatives (5a-g)

Compound	S. aureus	B. subtilis	E. coli	K. aerogenes
5a	++	++	-	+
5b	+++	+++	-	+++
5c	++	++	-	+
5d	+++	+++	-	++
5e	+++	++	-	++
5f	+++	++	-	++
5g	++++	+++	-	++
Amoxicillin	++++	++++	++++	++++

Zone of inhibition: (-) 0-5 mm, inactive; (+) 6-10 mm, poor activity; (+ +) 11-15 mm, moderate activity; (+ + +) 16-20 mm, good activity; (+ + + +) >20 mm, very good activity.

CONCLUSION

We herein report a microwave assisted method for the synthesis of cyclohexenone and indazole derivatives. This method is simple, efficient, high yielding as compared to conventional method and avoids the use of harmful solvents.

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REFERENCES

1. Abouzid KA, El-Abahr HS. Synthesis and anti-inflammatory activity of novel indazolones. Arch Pharm Res 2003; 26: 1-8.
2. Yakaiah T, Lingaiah BPV, Narsaiah B, Shireesha B, Ashok Kumar B, Gururaj S, Parthasarathyc T, Sridhare B. Synthesis and structure-activity relationships of novel pyrimido[1,2-b]indazoles as potential anticancer agents against A-549 cell lines. Bioorg. Med. Chem. Lett. 2007; 17: 3445-3453.
3. Alisi MA, Cazzolla N, Furlotti G, Guglielmotti A, Polenzani L US Pat. (2010); US 7,662,836 B2.
4. Rosati O, Curini M, Marcotullio MC, Macchiarulo A, Perfumi M, Mattioli L, Rismondob F, Cravottoc G. Synthesis, docking studies and anti-inflammatory activity of 4,5,6,7- tetrahydro-2H-indazole derivatives. Bioorg. Med. Chem. 2007; 15: 3463-3473.

5. Buchel GE, Stepanenko IN, Hejl M, Jakupec MA, Keppler BK, Heffeter P, Berger W, Arion VB. Osmium(IV) complexes with 1H- and 2H-indazoles: Tautomer identity versus spectroscopic properties and antiproliferative activity. *J. Inorg. Biochem.* 2012; 113: 47-54.
6. Raffa D, Maggio B, Cascioferro S, Raimondi MV, Schillaci D, Gallo G, Daidone G, Plescia S, Meneghetti F, Bombieri G, Cristina AD, Pipitone RM, Grimaudo S, Tolomeo M. Synthesis and antiproliferative activity of 3-amino-N-phenyl-1H-indazole-1-carboxamides. *Eur J Med Chem.* 2009; 44: 165-178.
7. May JA, Dantanarayana AP, Zinke PW, McLaughlin MA, Sharif NA. 1-((S)-2-aminopropyl)-1H-indazol-6-ol: a potent peripherally acting 5-HT₂ receptor agonist with ocular hypotensive activity. *J Med Chem.* 2006; 49: 318-328.
8. Vyas DH, Tala SD, Akbari JD, Dhaduk MF, Joshi HS. Synthesis, antimicrobial and antitubercular activity of some cyclohexenone and indazole derivatives. *Indian J Chem.* 2009; 48B: 1405-1410.
9. Ko JH, Yeon SW, Ryu JS, Kim TY, Song EH, You HJ, Park RE, Ryu CK. Synthesis and biological evaluation of 5-arylamino-6-chloro-1H-indazole-4,7-diones as inhibitors of protein kinase B/Akt. *Bioorg Med Chem Lett.* 2006; 16: 6001-6005.
10. Huang LJ, Shih ML, Chen HS, Pan SL, Teng CM, Lee FY, Kuo SC. Synthesis of N₂-(substituted benzyl)-3-(4-methylphenyl) indazoles as novel anti-angiogenic agents. *Bioorg Med Chem.* 2006; 14: 528-536.
11. Gautam D, Chaudhary RP. Synthesis, structure and antimicrobial evaluation of new 3,3a,4,5-tetrahydro-2H-benzo[g]indazol-2-yl-thiazol-4(5H)-ones. *Spectrochim Acta Part A: Molecular and Biomol Spect.* 2015; 135: 219-226.
12. Minu M, Thangadurai A, Wakode SR, Agrawal SS, Narasimhan B. Synthesis, antimicrobial activity and QSAR studies of new 2,3-disubstituted-3,3a,4,5,6,7-hexahydro-2H-indazoles. *Bioorg Med Chem Lett.* 2009; 19: 2960-2964.
13. Raffa D, Daidone G, Plescia F, Schillaci D, Maggio B, Torta L. Synthesis and antifungal activity of new N-(1-phenyl-4-carboxypyrazol-5-yl)-, N-(indazol-3-yl)- and N-(indazol-5-yl)-2-iodobenzamides. *Farmaco.* 2002; 57: 183-187.
14. Cotterill IC, Usyatinsky AY, Arnold JM, Clark DS, Dordick JS, Michels PC, Khmel'nitsky YL. Microwave Assisted Combinatorial Chemistry Synthesis of Substituted Pyridines. *Tetrahedron Lett.* 1998; 39: 1117-1120.
15. Deshayes S, Liagre M, Loupy A, Luche JL, Petit A. Microwave Activation in Phase Transfer Catalysis. *Tetrahedron.* 1999; 55: 10851-10870.

16. Chandekar CJ, Madhugiri MJ. Antimicrobial Potential of Leaves of Psidium Guajava. The Bioscan. 2011; 6(4): 557-561.

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